

IN THE SPECIFICATION

Please replace the second full paragraph on page 10 of the Specification with the following amended paragraph:

The internal stability of sodium percarbonate can be represented as a TAM value, according to which the stability increases with decreasing TAM value. A good storage life is indicated by a low TAM value. The TAM value should preferably be below 10 $\mu\text{W/g}$ and in particular below 8 $\mu\text{W/g}$ for sodium percarbonate. The TAM value is a microcalorimetric determination of the energy released during storage, measured by means of the TAM® Thermal Activity Monitor from Thermometric AB, Järfälla (Sweden). The TAM features four measuring cylinders, each containing a pair of measuring cups sandwiched between a pair of Peltier thermopile heat sensors. The cylinders are surrounded by an infinite heat sink: a water bath that can be maintained at a temperature. The thermopile heat sensor elements of the two measuring cups in each cylinder are connected in series but in opposition, so that the resultant signal represents the difference in heat flow from the ~~two~~ two cups. One contains the sample and the other contains a reference. As the sodium percarbonate degrades, it gives off heat and tries to restore thermal equilibrium with the waterbath. The flow of this heat is detected, measured, and converted to a TAM value in $\mu\text{W/g}$.